

Predicting dietborne metal toxicity from metal influxes

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4 pages (2 Tables and 4 equations)

Table S1. Nominal dissolved tracer concentrations ($\mu\text{g l}^{-1}$) used to label *N. palea* offered as food to *L. stagnalis*

Experiment	^{65}Cu	^{106}Cd	^{62}Ni
1	10	1.0	2.0
1	100	10	20
1	250	25	50
1	500	50	100
1	1000	100	200
2	0	2.0	4.0
2	0	10	20
2	0	20	40
2	0	50	100
2	0	100	200
2	0	200	400
3	20	0	0
3	100	0	0
3	200	0	0
3	400	0	0
3	800	0	0
3	1200	0	0

Calculation of accumulated ^{106}Cd concentrations. The relative abundance of ^{106}Cd is determined using the net signal intensities of each Cd isotope in the standards used to calibrate the ICP-MS. For example,

$$p^{106} = \text{Intensity} \left(\frac{^{106}\text{Cd}}{^{106}\text{Cd} + ^{108}\text{Cd} + ^{110}\text{Cd} + ^{111}\text{Cd} + ^{112}\text{Cd} + ^{113}\text{Cd} + ^{114}\text{Cd}} \right)_{\text{Standard}} \quad (1)$$

Concentrations of ^{106}Cd in the experimental organisms ($[^{106}\text{Cd}]_e$) are calculated as the product of p^{106} and the total Cd concentrations inferred by the ICP-MS software from tracer intensity ($[T^{106}\text{Cd}]$):

$$[^{106}\text{Cd}]_e = p^{106} \times [T^{106}\text{Cd}] \quad (2)$$

Total Cd concentrations inferred from the intensity of the most abundant isotope are then used to derive the original load of tracer ($[^{106}\text{Cd}]_e^0$) that occurred in each sample in the absence of a spike, e.g.,

$$[^{106}\text{Cd}]_e^0 = p^{106} \times [T^{114}\text{Cd}] \quad (3)$$

Net ^{106}Cd uptake ($\Delta[^{106}\text{Cd}]_e$) is $[^{106}\text{Cd}]_e$ as derived from the total experimental Cd inferred from ^{106}Cd signal (equation 2) minus the pre-existing load of ^{106}Cd ($[^{106}\text{Cd}]_e^0$) from equation 3,

$$\Delta[^{106}\text{Cd}]_e = [^{106}\text{Cd}]_e - [^{106}\text{Cd}]_e^0 \quad (4)$$

TABLE S2. Food IR ($\text{g g}^{-1} \text{ d}^{-1} \pm 95\% \text{ C.I.}$) determined by mass-balance calculations for each tracer

Metal exposure (nmol g^{-1})	^{65}Cu	^{106}Cd	^{62}Ni
Cu + Cd + Ni			
1373 $\pm$ 240	0.136 $\pm$ 0.039 (6)	0.189 $\pm$ 0.039 (8)	0.050 $\pm$ 0.024 (7)
10718 $\pm$ 354	0.118 $\pm$ 0.041 (8)	0.115 $\pm$ 0.040 (8)	0.058 $\pm$ 0.022 (8)
20324 $\pm$ 922	0.097 $\pm$ 0.027 (8)	0.087 $\pm$ 0.022 (8)	0.047 $\pm$ 0.015 (8)
32946 $\pm$ 7154	0.078 $\pm$ 0.023 (8)	0.050 $\pm$ 0.015 (8)	0.020 $\pm$ 0.007 (8)
59923 $\pm$ 6813	0.045 $\pm$ 0.015 (7)	0.015 $\pm$ 0.003 (7)	0.004 $\pm$ 0.002 (7)
Cd + Ni			
263 $\pm$ 23		0.152 $\pm$ 0.070 (10)	0.158 $\pm$ 0.069 (9)
1449 $\pm$ 69		0.168 $\pm$ 0.059 (10)	0.131 $\pm$ 0.042 (10)
2369 $\pm$ 434		0.222 $\pm$ 0.096 (9)	0.164 $\pm$ 0.076 (9)
4940 $\pm$ 622		0.236 $\pm$ 0.084 (10)	0.158 $\pm$ 0.055 (10)
8073 $\pm$ 632		0.256 $\pm$ 0.088 (10)	0.148 $\pm$ 0.041 (10)
11339 $\pm$ 399		0.303 $\pm$ 0.100 (10)	0.208 $\pm$ 0.075 (10)
Cu only			
1420 $\pm$ 83	0.091 $\pm$ 0.037 (8)		
9808 $\pm$ 744	0.138 $\pm$ 0.073 (7)		
20523 $\pm$ 1052	0.111 $\pm$ 0.025 (10)		
33847 $\pm$ 757	0.068 $\pm$ 0.019 (8)		
83848 $\pm$ 6977	0.017 $\pm$ 0.004 (10)		
117057 $\pm$ 14377	0.008 $\pm$ 0.003 (10)		